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Publisher: Taylor & Francis

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Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gmcl16

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To cite this article: Virgil Percec (1984): Microstructure of Polyphenylacetylene and Other Acetylenic

Polymers, Molecular Crystals and Liquid Crystals, 106:3-4, 409-409

To link to this article: http://dx.doi.org/10.1080/00268948408071462

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Mol. Cryst. Liq. Cryst., 1984, Vol. 106, p. 409 0026-8941/84/1064-0409/\$18.50/0 © 1984 Gordon and Breach, Science Publishers, Inc Printed in the United States of America

MICROSTRUCTURE OF POLYPHENYLACETYLENE AND OTHER ACETYLENIC POLYMERS

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Abstract. ¹H- and ¹³C-NMR studies carried out on poly(phenylacetylene)s and poly(pentadeuterophenylacetylene)s prepared with Ziegler and metathesis type catalysts, revealed two different mechanisms which control the microstructure of acetylenic polymers.

For the Ziegler-type catalysts, thermal induced reactions (intramolecular cyclization, aromatization, chain scission and cis-trans isomerization) of the cis-double bonds (isomerization after double bond formation) are responsible for the polymer microstructure. (For a review see: Progr. Polym. Sci., 8, 133 (1982).) Both isomerization "after" double bond formation as well as isomerization "before" double bond formation are responsible for the microstructure of polymers obtained with metathesis type catalysts. In both cases, the polymer molecular weight is controlled by the intramolecular cyclization of the cis-polymer chain end. Under reaction conditions where isomerization occurs mainly "before" double bond formation, "pure" trans polymers having very high molecular weight are obtained (Polym. Prepr., 24(1), 239 (1983); Polym. Bull., 9, 548 (1983)).

The microstructure of several other acetylenic polymers will be commented on in relation with the polymerization mechanism and with the microstructure of (CH).